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# ProductInformation

Folic acid Cell Culture Tested

Product Number **F 8758** Store at Room Temperature

### **Product Description**

Molecular Formula:  $C_{19}H_{19}N_7O_6$ Molecular Weight: 441.4 CAS Number: 59-30-3  $pK_a$ : 8.26<sup>1</sup>  $\lambda_{max}$ : 368, 283, 256 nm<sup>2</sup> Extinction Coefficient:  $E^{mM}$ = 9.12, 25.1, 26.9 (0.1 M NaOH)<sup>2</sup> Specific Rotation: +23° ( 0.5 %, 0.1 M NaOH, 25 °C) Synonym: pteroylglutamic acid<sup>3</sup>

Folic acid, also known as folate, is a B vitamin that can be found in a variety of fruits and vegetables. It can also be chemically synthesized. Folate, a watersoluble vitamin, helps the body form red blood cells and aids in the formation of genetic material within every body cell.

This product exhibits metal binding properties.<sup>1</sup> The log stability constants for various cations are as follows:

6.0 (Mn<sup>2+</sup>) 7.9 (Fe<sup>2+</sup>) 8.1 (Co<sup>2+</sup>) 9.0 (Ni<sup>2+</sup>) 7.8 (Cu<sup>2+</sup>) 7.5 (Zn<sup>2+</sup>)

This product is synthetic and is 100% of the L-isomer. It is tested for use in cell culture applications at 4 mg/L on both mammalian cells and insect cell lines.

## **Precautions and Disclaimer**

For Laboratory Use Only. Not for drug, household or other uses.

### **Preparation Instructions**

This product is soluble in 1 M NaOH (50 mg/ml). The free acid is only slightly soluble in water (0.01 mg/ml at 0  $^{\circ}$ C) and is insoluble in aqueous solutions below pH 5. It is soluble in sodium bicarbonate or sodium hydroxide solutions.

### Storage/Stability

This product is photolabile and is inactivated by UV light.<sup>3</sup> Alkaline solutions are reasonably stable in the dark, but heating will destroy both alkaline and acidic solutions.

#### References

- Data for Biochemical Research, 3rd ed., Dawson, R. M. C., et al., Oxford University Press (New York, NY: 1986), p. 412.
- 2. The Merck Index, 11th ed., Entry# 4140.
- Data for Biochemical Research, 3rd ed., Dawson, R. M. C., et al., Oxford University Press (New York, NY: 1986), p. 134-135.

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